FLINN SCIENTIFIC, INC. Material Safety Data Sheet (MSDS)

MSDS #: 56.00

Revision Date: December 8, 2011

SECTION 1 — CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

Ammonium Dichromate

Flinn Scientific, Inc. P.O. Box 219 Batavia, IL 60510 (800) 452-1261

CHEMTREC Emergency Phone Number: (800) 424-9300

SECTION 2 — COMPOSITION, INFORMATION ON INGREDIENTS

Ammonium dichromate

Synonym: Ammonium bichromate

CAS#: 7789-09-5

SECTION 3 — HAZARDS IDENTIFICATION

Orange crystals. Odorless.

Known human carcinogen (IARC-1).

Strong oxidizing agent and dangerous fire risk if mixed with organic material. Ingestion can be fatal. Inhalation causes severe irritation. Avoid inhaling dust. Irritating to body tissues. Avoid all body tissue

FLINN AT-A-GLANCE

Health-3

Flammability-1

Reactivity-3

Exposure-3

Storage-0

0 is low hazard, 3 is high hazard

SECTION 4 — FIRST AID MEASURES

Call a physician and seek medical attention for further treatment, observation, and support after first aid. Inhalation: Remove to fresh air at once. If breathing has stopped, give artificial respiration immediately.

Eye or External: Immediately flush with fresh water for at least 15 minutes.

Internal: Rinse out mouth, give 1 to 2 cups of water or milk, induce vomiting. Call a physician or poison control at once.

SECTION 5 — FIRE FIGHTING MEASURES

Strong oxidizer under fire conditions. Dangerous fire risk.	NFPA CODE
Thermal decomposition of ammonium dichromate is initiated by local heating to 190 °C. Flame and sparks	H-2
spread rapidly through the mass, and if confined, it may become explosive.	F-1
Fire Fighting Instructions: Use a triclass, dry chemical fire extinguisher. Firefighters should wear PPE and	R-1
SCBA with full facepiece operated in positive pressure mode.	OX

SECTION 6 — ACCIDENTAL RELEASE MEASURES

Restrict unprotected personnel from the area and ventilate area. Sweep up the spill, place in a sealed bag or container, and dispose. Ventilate area and wash spill site after material pickup is complete. See Sections 8 and 13 for further information.

SECTION 7 — HANDLING AND STORAGE

Flinn Suggested Chemical Storage Pattern: Inorganic #8. Store with borates, chromates, manganates, and permanganates. Use and dispense in a hood.

SECTION 8 — EXPOSURE CONTROLS, PERSONAL PROTECTION

Avoid contact with eyes, skin, and clothing. Wear chemical splash goggles, chemical-resistant gloves, and chemical-resistant apron. Use ventilation to keep airborne concentrations below exposure limits. Always wear a NIOSH-approved respirator with proper cartridges or a positive-pressure, air-supplied respirator when handling this material in emergency situations (spill or fire). Exposure guidelines: TWA 0.05 mg/m³ (as Cr) (ACGIH)

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SECTION 9 — PHYSICAL AND CHEMICAL PROPERTIES

Orange crystals.

Solubility: Water soluble (30%).

Formula: (NH₄)₂Cr₂O₇ Formula weight: 252.06 Melting point: 170 °C (decomposes)

Specific gravity: 2.15

SECTION 10 — STABILITY AND REACTIVITY

Strong oxidizer. Avoid contact with organic or easily oxidized substances. Avoid sources of heat. Avoid reducing agents, organic substances, and oxidizing agents.

Shelf life: Fair to poor.

SECTION 11 — TOXICOLOGICAL INFORMATION

Acute effects: Poison, irritant, sensitizer, stomach pains, vomiting,

and diarrhea.

Chronic effects: Carcinogen; possible mutagen.

Target organs: Lungs, kidneys.

ORL-RAT LD₅₀: N.A.

IHL-RAT LC₅₀: N.A.

SKN-RBT LD₅₀: N.A.

N.A. = Not available, not all health aspects of this substance have been fully investigated.

SECTION 12 — ECOLOGICAL INFORMATION

Data not yet available.

SECTION 13 — DISPOSAL CONSIDERATIONS

Please review all federal, state and local regulations that may apply, before proceeding.

Flinn Suggested Disposal Method #12a is one option.

SECTION 14 — TRANSPORT INFORMATION

Shipping name: Ammonium Dichromate

Hazard class: 5.1, Oxidizer UN number: UN1439 N/A = Not applicable

SECTION 15 — REGULATORY INFORMATION

TSCA-listed, EINECS-listed (232-143-1), RCRA code D001, D007.

SECTION 16 — OTHER INFORMATION

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Consult your copy of the Flinn Science Catalog/Reference Manual for additional information about laboratory chemicals.