

# MATERIAL SAFETY DATA SHEET

BB0055

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MSDS No .

Approved by:

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MSDS No.: BB0055	
Section 1	Chemical Product and Company Information
Product	BARIUMNITRATE
Synonyms	Barium Dinitrate
CHEMTREC 24 Hour Emergency Phone Number (800) 424-9300	

#### Section 2 Hazards Identification

**Emergency Overview** 

FATAL IF SWALLOWED.

# DANGER! STRONG OXIDIZER! POISON &

= Minimal	Health	3	
= Slight	Fire	0	
= Moderate = Serious	Reactivity	3	
= Severe	Contact	2	
	HMIS	*	

Keep from contact with clothing and other combustible materials. Keep in a cool, dry place. Avoid contact with skin and eyes. Wash thoroughly after handling. Target organs: Central nervous system, kidneys.

CONTACT WITH OTHER MATERIAL MAY CAUSE FIRE OR EXPLOSION. MAY BE

Section 3 Composition / Information on Ingredients				
Chemical Name		CAS #	%	TLV Units
Barium nitrate		10022-31-8	100%	TWA: 0.5 mg/m <sup>3</sup> (as Barium) (ACGIH 2001)
Section 4	First Aid Massures		,	•

#### Section 4 First Aid Measures

INGESTION: Call physician or Poison Control Center immediately. Induce vomiting only if advised by appropriate medical personnel. Never give anything by mouth to an unconscious person.

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

EYE CONTACT: Check for and remove contact lenses. Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention.

SKIN CONTACT: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get medical attention.

# Section 5 **Fire Fighting Measures**

General information: In fire conditions, wear a NIOSH/MSHA-approved self-contained breathing apparatus and full protective gear. Containers may rupture when involved in fire. Use water spray to keep fire-exposed containers cool. Although not flammable, substance is a strong oxidizer which releases oxygen on heating, increasing the burning rate of any material.

Extinguishing Media: Water spray, carbon dioxide, dry chemical. Flash Point: Non-flammable. Autoignition temperature: N/A Explosion Limits: Lower: N/A Upper: N/A	0 = Minimal 1 = Slight 2 = Moderate 3 = Serious 4 = Severe
Section 6 Accidental Release Measures	
Use proper personal protective equipment as indicated in Section 8. Provide adequate v	entilation. Recover for

use if not contaminated. Sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways.

# Section 7 Handling & Storage

**OXIDIZER STORAGE CODE YELLOW** 

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children.

Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale dusts. Wash thoroughly after handling. Remove and wash clothing before reuse.

Storage: Store in a cool, dry, well-ventilated area away from easily oxidizable substances.

#### **Exposure Controls / Personal Protection** Section 8

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield. lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: None should be needed in normal laboratory handling at room temperatures. If dusty conditions prevail, work in fume hood or wear a NIOSH/MSHA-approved respirator.

Section 9 Physical & Chemical Properties
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Physical state: Solid.	Boiling point: Decomposes.
Appearance: White crystalline powder.	Freezing / Melting point: 592°C (1097°F)
Odor: No odor.	Decomposition temperature: N/A
pH: N/A	Solubility: 8.7 g/100ml water @ 20°C
Vapor pressure (mm Hg): Negligible.	Specific gravity (H <sub>2</sub> O = 1): 3.24 @ 23°C
Vapor Density (Air = 1): N/A	Percent volatile (%): Negligible.
Evaporation rate (Butyl acetate = 1): N/A	Molecular formula: Ba(NO <sub>3</sub> ) <sub>2</sub>
Viscosity: N/A	Molecular weight: 261.35
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Section 10 Stability & Reactivity	

Chemical stability: Stable Hazardous polymerization: Will not occur. Conditions to avoid: Do not heat or rub with organic matter or other oxidizable substance. e.g. sulfur, sulfides, phosphides, hypophosphites, etc.

Incompatibilities with other materials: Barium oxides, magnesium and zinc, reducing agents and combustible materials

Hazardous decomposition products: Nitrogen oxides, barium oxide, barium dust and/or fume.

#### Section 11 **Toxicological Information**

Effects of overexposure: Contact with skin and eyes may cause irritation, redness and/or pain. Ingestion may cause weakness, salivation and nausea, followed by vomiting and diarrhea. Patient may become cold and experience varying degrees of paralysis. May be harmful by inhalation, with symptoms similar to those of ingestion. Exercise appropriate procedures to minimize potential hazards.

RTECS #: CQ9625000 ORAL-RAT LD50: 355 mg/kg

# Section 12 **Ecological Information**

Data not yet available.

# Section 13 **Disposal Considerations**

These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local state and federal regulations or contract with a licensed chemical disposal agency.

Section 14	Transport Information
JN/NA number:	UN1446
Shipping name:	Barium nitrate
Hazard class: 5	.1, (6.1)
Packing group:	II
Exceptions: Lim	nited quantity equal to or less than 0.5 Kg (1.1 lb).
Section 15	Regulatory Information

TSCA-listed, EINECS-listed (233-020-5), RCRA-Code D001, D005, DSL-listed.

#### Section 16 Additional Information

The information contained herein is furnished without warranty of any kind. Employers should use this information only as a supplement to other information gathered by them and must make independent determinations of suitability and completeness of information from all sources to assure proper use of these materials and the safety and health of employees, \* Hazardous Materials Industrial Standards.